







SOCIÉTÉ DE GESTION DES DÉCHETS NUCLÉAIRES



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#### **NWMO: Who We Are**

- Formed in 2002 as required by *Nuclear Fuel Waste Act*
- Funded by Canada's nuclear energy corporations
- Operates on a not-for-profit basis

Our mission is to develop and implement collaboratively with Canadians, a management approach for the long-term care of Canada's used nuclear fuel that is socially acceptable, technically sound, environmentally responsible, and economically feasible.



### Site Selection Process: Initiated May 2010

## Seeking an informed and willing host with a suitable geologic formation:

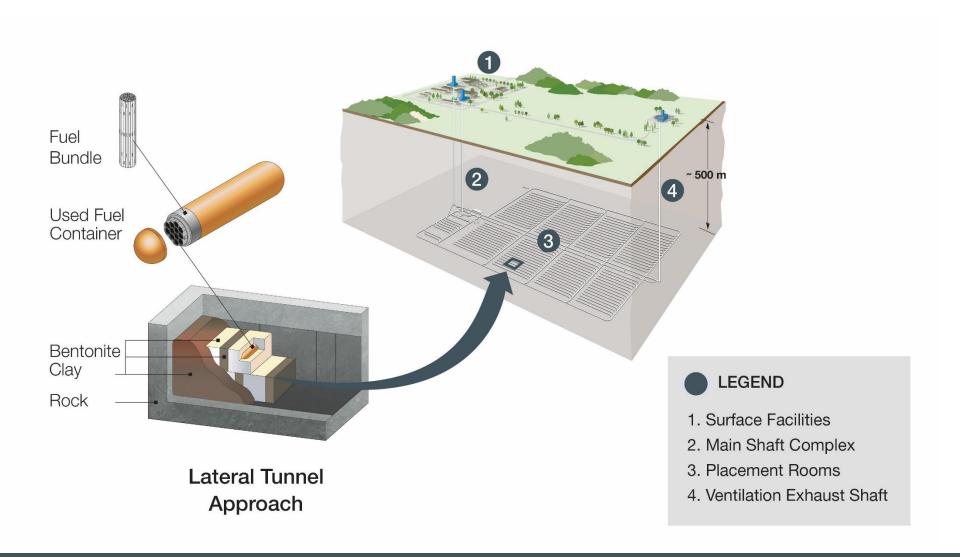
- Multi-stage technical and socio-economic assessment approach
- Phased process over many years
- Communities expressed interest to participate
- Communities can choose to leave the process

# Centralized containment and isolation of used nuclear fuel in a deep geological repository

The project will only proceed with interested community, First Nation and Métis communities and surrounding municipalities working in partnership.

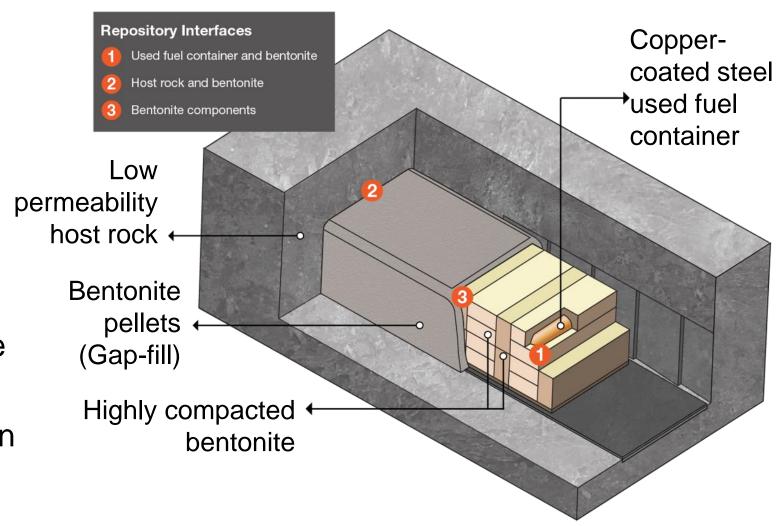


### Deep Geological Repository (DGR)



### Natural and Engineered Barriers in a DGR

- Multiple barriers between used fuel and the environment including:
  - Host rock
  - Bentonite
  - UFC (copper coated)
- Copper is a naturally stable metal but can be susceptible to sulphide corrosion
- Sulphate reducing bacteria in the host rock could produce sulphide.



### **Sulphide Transport Modelling**

- Steady state, isothermal analysis
- Transient, thermal analysis (1 million years)
- Sensitivity analysis of:
  - Concentration gradient
  - Diffusion coefficient
  - Buffer box size (bentonite thickness)
  - Preferential pathways (hot spots, fractures, bentonite failure)
- The model was developed to aid in the safety assessment of the copper coating



### **Methods and Assumptions**

Sulphide Production:

$$2Cu(s) + HS^- + H^+ \rightleftarrows Cu_2S(s) + H_2$$

Fick's Law of Diffusion:

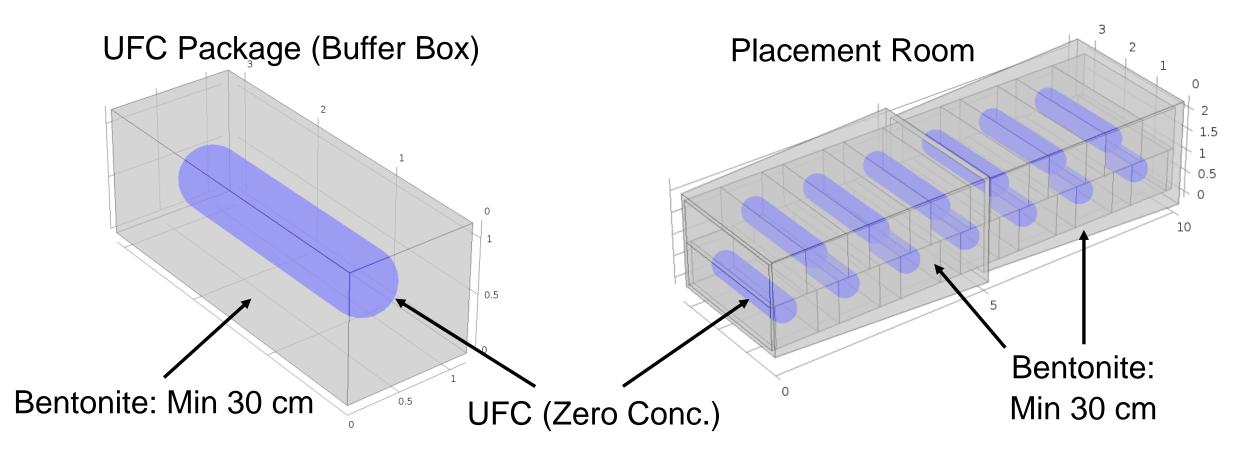
$$J = -D\nabla C$$

Corrosion Depth:

$$d_{corr} = \frac{N_{HS} f_{HS} M_{cu}}{A_{corr} \rho_{cu}}$$

It is assumed that diffusion is the only process taking place (no active SRB, geochemistry, advective flux ...) in a homogenous bentonite.

### **Model Geometry and Boundary Conditions**

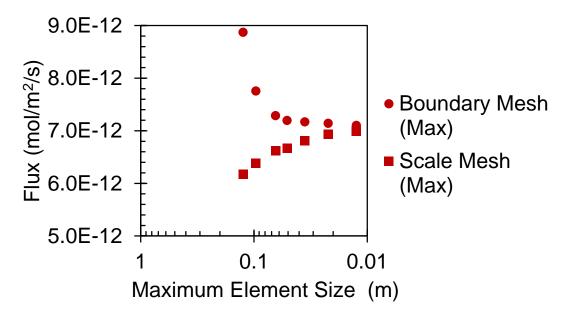


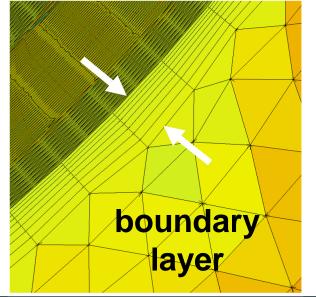
Constant sulphide concentration surrounding each model geometry

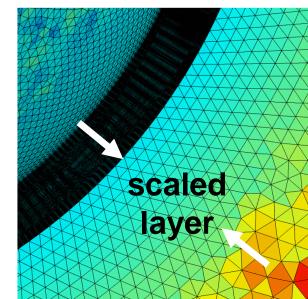
#### **Model Implementation**

#### Mesh Development:

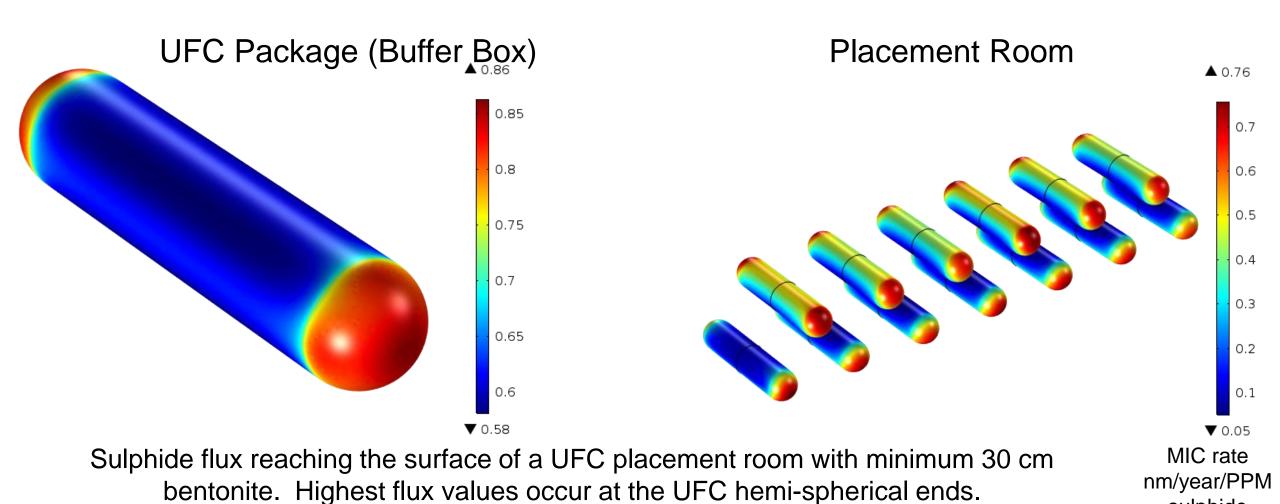
- Compared two mesh types
- Ensured a mesh size independent solution.
- Single UFC: 18.7 million elements (max: 0.015m)
- Placement Room: 20.6 million elements (max: 0.03m)
- Time dependent solve time ~ 1 day







### **Steady State Sulphide Transport**

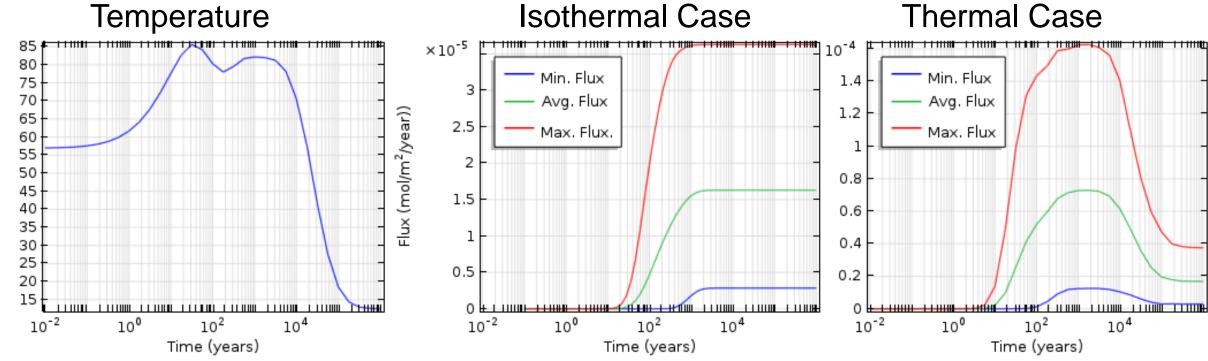


sulphide

### **Steady State Sulphide Transport**

Model Type	MIC Rate (mm per 1 million years per PPM sulphide)
3D – UFC package – minimum <i>20 cm</i> bentonite	0.8 (average), 1.1 (max.)
3D – UFC package – minimum <i>30 cm</i> bentonite	0.65 (average), 0.86 (max.)
3D – Placement room – minimum 30 cm bentonite	0.31 (average), 0.76 (max.)

### **Thermal Sulphide Modelling**

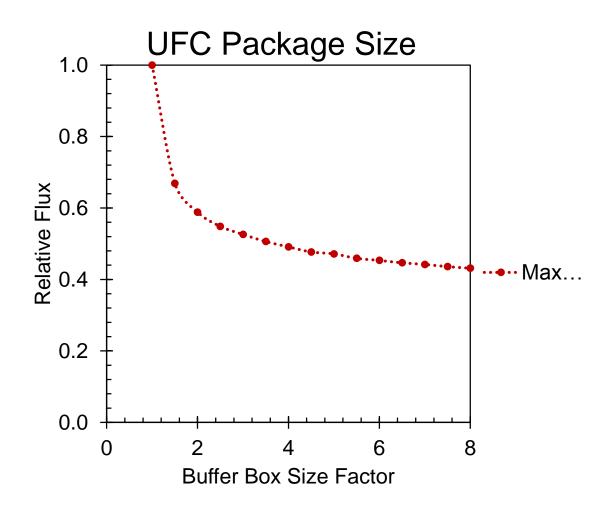


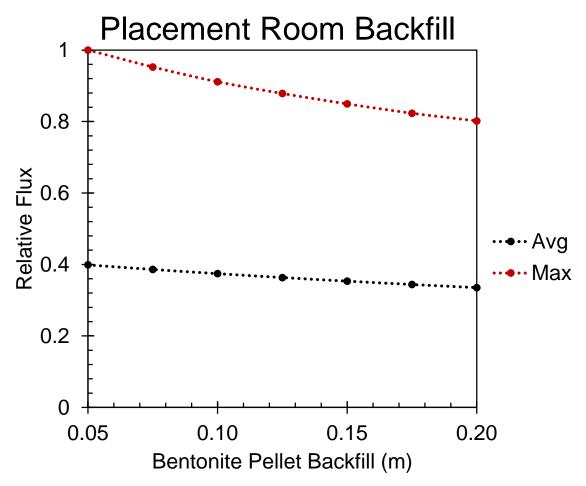
#### Maximum corrosion rates:

- Isothermal case at 25°C: 0.77 mm/1 million years/ppm sulphide
- Isothermal case at 11°C: 0.52 mm/1 million years/ppm sulphide
- Thermal case (cumulative): 0.60 mm total over 1 million years (per ppm sulphide)



#### **Sensitivity Analysis – Bentonite Thickness**





#### **Sulphide Transport Modelling - Summary**

- Steady state isothermal and transient thermal analysis
- Sensitivity analysis of:
  - Concentration gradient linear dependence
  - Diffusion coefficient linear dependence
  - Buffer box size (bentonite thickness) non-linear dependence
- Localized sulphide fluxes may be higher than average rates or those predicted by 1D models due to the UFC geometry and the inclusion of 3D diffusive transport modelling.
- The developed model can aid in the safety assessment of the UFC copper coating.



### Thank you

#### Acknowledgements:

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#### **Extra Slides**



#### **Effective Diffusion Coefficient**

•  $D_{eff}$  is the effective diffusion coefficient at 25°C: 1x10<sup>-11</sup> m<sup>2</sup>/s  $D_{eff} = \tau \theta_{eff} D_w$ 

#### where ....

- $D_w$  is the diffusion coefficient of sulphide in water (Chemistry Handbook, 2009),
- The tortuosity  $(\tau)$  is assumed to 0.1 (Oscarson et al. 1994) and
- Total porosity is assumed to be a sum of accessible and non-accessible pores and the
  accessible pores are further broken into transport porosity and storage porosity. We assume an
  transport porosity = 0.1