

Modeling Dielectric Heating: A First Principles Approach

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Pryor Knowledge Systems

Introduction: Dielectric Heating: A First Principles Approach

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What are the Two Types of Electromagnetic Heating?

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1. Joule Heating

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2. Dielectric Heating

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How are the Two Types of Electromagnetic Heating Different?

Dielectric Heating: A First Principles Approach

How are the Two Types of Electromagnetic Heating Different?

1. Joule Heating

Based on Ohm's Law

How are the Two Types of Electromagnetic Heating Different?

1. Joule Heating

<i>Based on Ohm's Law</i>	<i>Power (Watts/second)</i>
<i>Applied over time =</i>	$= I(\text{current}) * V(\text{voltage})$
<i>Joule's First Law</i>	$= I^2(\text{current}^2) * R(\text{resistance})$
	$= \frac{V^2(\text{voltage}^2)}{R(\text{resistance})}$

How are the Two Types of Electromagnetic Heating Different?

1. Joule Heating

Based on Ohm's Law Mechanism: Charged Free Carrier Flow
Applied over time = Electrons, Electrons and/or Ions
Joule's First Law

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

*Non-Ohmic Heating
Mechanism*

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

*Non-Ohmic Heating
Mechanism*

*First discovered and patented at
Bell Telephone Laboratories
in 1937 by Joseph G. Chaffee*

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

*Non-Ohmic Heating
Mechanism*

*Dielectric Heating typically occurs in
the absence of free carriers in
nominally non-conductive materials
(dielectrics).*

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

*Non-Ohmic Heating
Mechanism*

*Dielectric Heating occurs as a result
of the relative motion of bound
electrons, ions, atoms, and molecules,
in situ.*

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

*Non-Ohmic Heating
Mechanism*

During electromagnetic excitation, the relative motion (vibration) of the bound electrons, ions, atoms, and molecules, in situ, couple to the bulk of the material. These vibrations are known as heat.

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

Non-Ohmic Heating Mechanism

ϵ_0 = permittivity of Free Space

c_0 = speed of light

μ_0 = permeability of Free Space

Different materials have a different relative permittivity (dielectric constant). The permittivity of Free Space is:

$$\epsilon_0 = \frac{1}{c_0 \mu_0} = 8.85419 \times 10^{-12} [F / m]$$

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

*Non-Ohmic Heating
Mechanism*

$$\epsilon = \epsilon_r \epsilon_0$$

where :

ϵ_r = relative permittivity

ϵ_0 = permittivity of Free Space

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

Non-Ohmic Heating

Mechanism

For very
complicated materials
with phase delays
and energy losses

$$\hat{\varepsilon} = \varepsilon'(\omega) + i\varepsilon''(\omega)$$

where :

$$\varepsilon'(\omega) = \text{real part}$$

$$\varepsilon''(\omega) = \text{imaginary part}$$

$$\omega = 2 * \pi * f_0 \text{ and } f_0 = \text{frequency}$$

How are the Two Types of Electromagnetic Heating Different?

2. Dielectric Heating

Non-Ohmic Heating Mechanism The loss equation is:

For very complicated materials

with phase delays and energy losses

$$Q_d = 2 * \pi * f_0 * \epsilon'' * \epsilon_0 * \left(\frac{V_{in}}{thickness} \right)$$

where : f_0 = frequency

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Dielectric Heating Model Materials

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Dielectric Heating Model Materials

whey gel = water + impurities ($\sim 10\%$)

formed into a gelatinous mass

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Dielectric Heating Model Materials

whhey gel = water + impurities (~10%)

formed into a gelatinous mass

Frequency (measured)	Real Part (measured)	Imaginary Part (measured)
27 MHz	96.30	1879.15
40 MHz	83.95	1263.20
915 MHz	54.20	61.30
1800 MHz	51.70	34.57

After: Yifen Wang, et.al., J. Food Eng., 57 (2003), pp. 257-268

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Dielectric Heating Model Materials

Real Part Fit Equation

$$\begin{aligned}\epsilon' &= 54.6919 - 2.05493 \times 10^{-9} * f_0 \\ &+ 1.27475 \times 10^9 * f_0^{-1} - 4.04542 \times 10^{15} * f_0^{-2} \\ f_0 &= \text{frequency}\end{aligned}$$

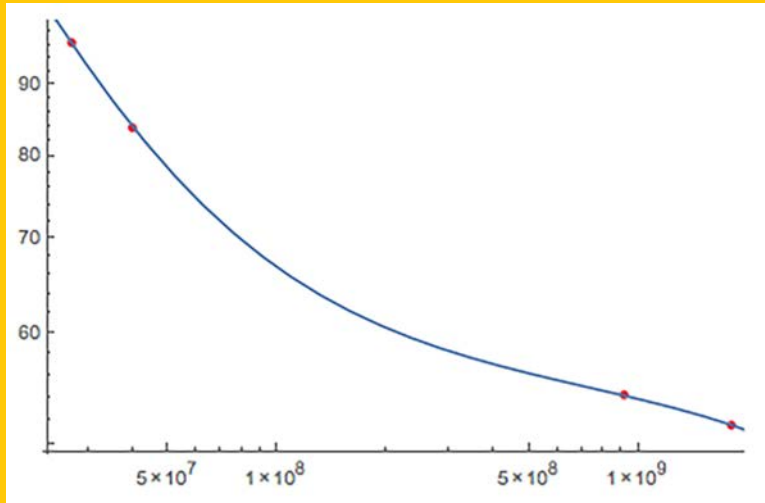
Imaginary Part Fit Equation

$$\begin{aligned}\epsilon'' &= 6.88131 + 1.59761 \times 10^{-10} * f_0 \\ &+ 4.96322 \times 10^{10} * f_0^{-1} + 2.48106 \times 10^{16} * f_0^{-2}\end{aligned}$$

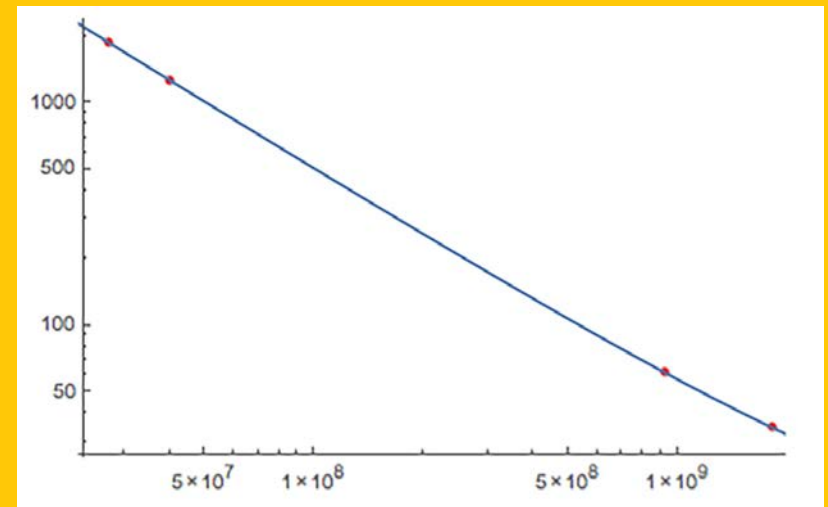
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Dielectric Heating Model Materials

Combined Real Part



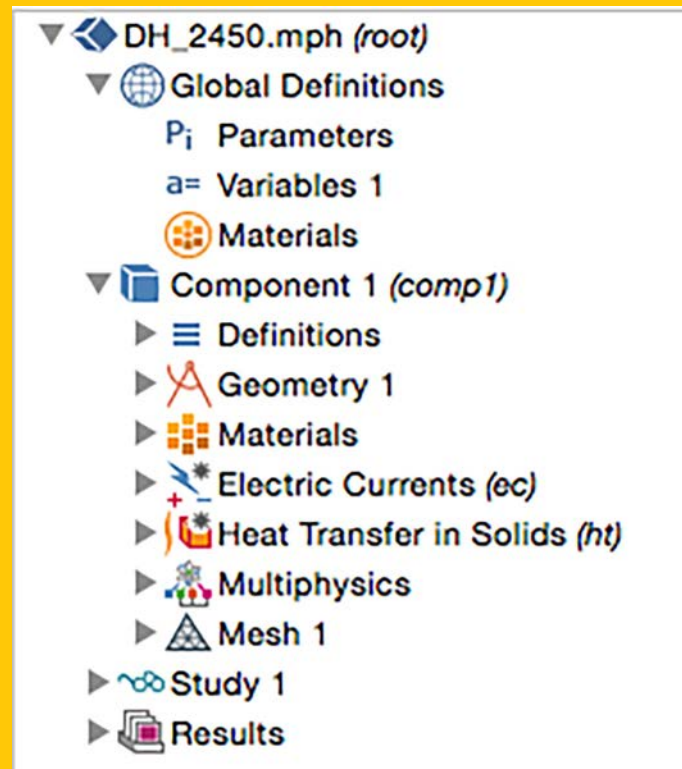
Combined Imaginary Part



Data and Fit Plots

Dielectric Heating: COMSOL Multiphysics Model

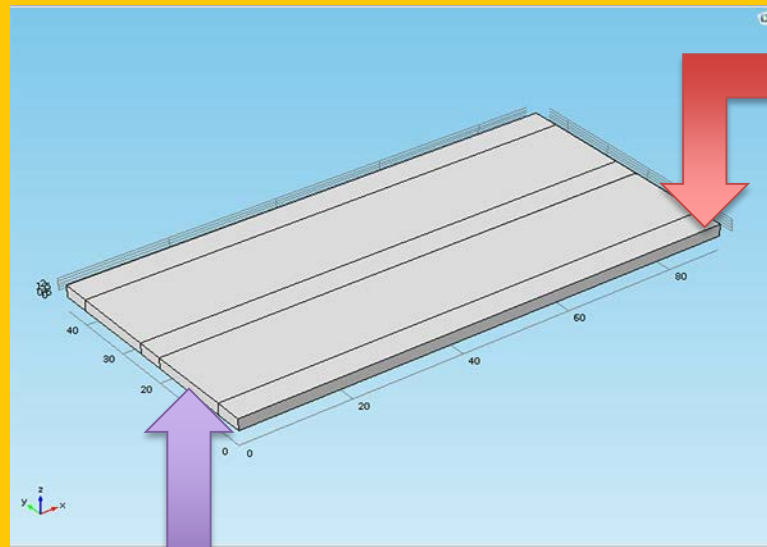
Model
Builder
Tree



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Dielectric Heating: COMSOL Multiphysics Model

Geometry →



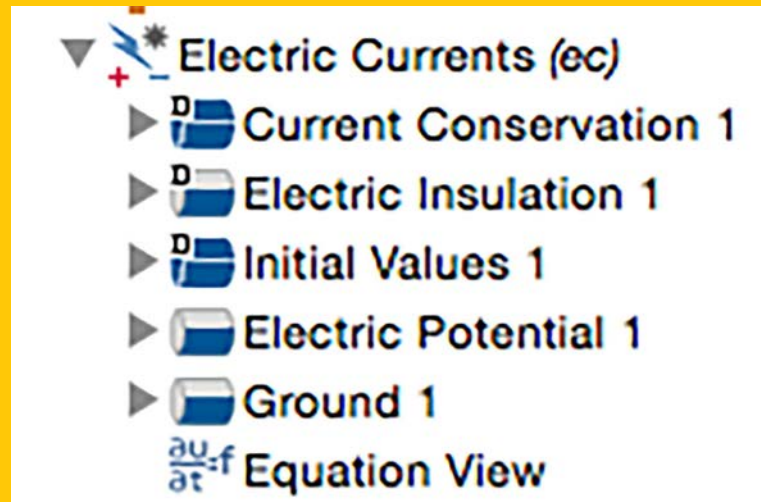
Air

Whey Gel

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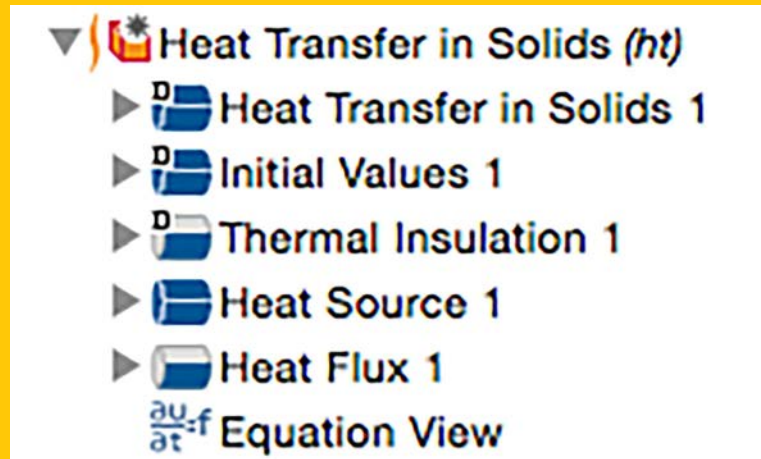
Dielectric Heating: COMSOL Multiphysics Model

Electric
Currents



Dielectric Heating: COMSOL Multiphysics Model

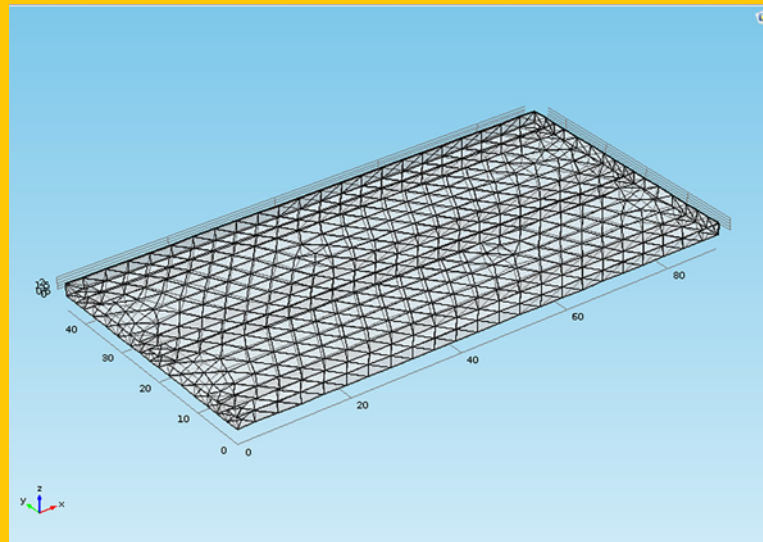
Heat
Transfer



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Dielectric Heating: Solving the COMSOL Multiphysics Model

**Model
Mesh**

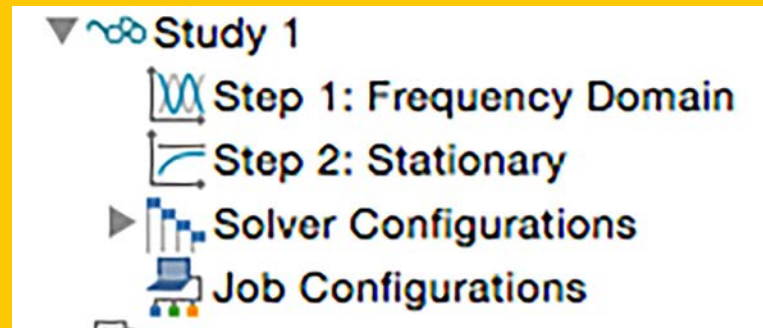


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Dielectric Heating: Solving the COMSOL Multiphysics Model

**Solver
Configuration** →

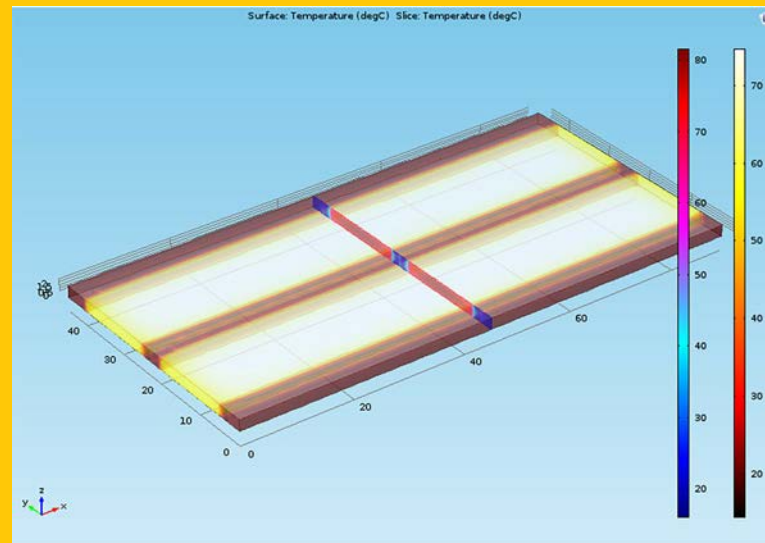


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Dielectric Heating: Solutions to the COMSOL Multiphysics Model

Solution
2.45 GHz

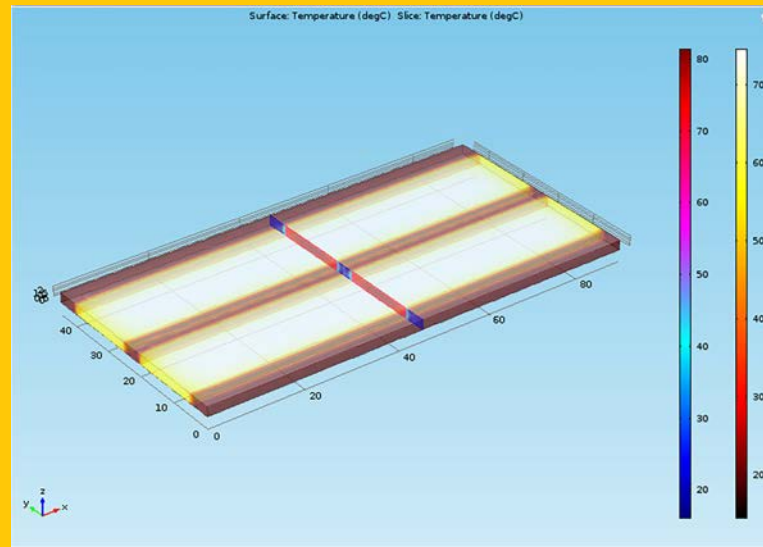


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Dielectric Heating: Solutions to the COMSOL Multiphysics Model

Solution
915 MHz

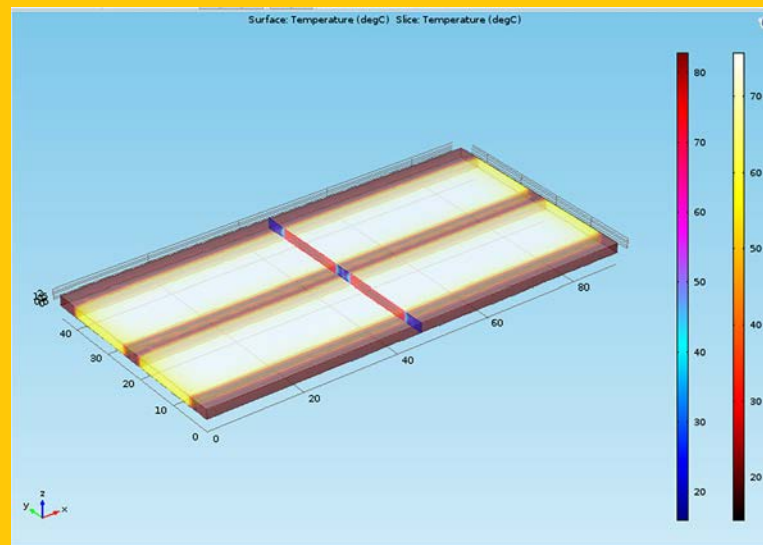


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Dielectric Heating: Solutions to the COMSOL Multiphysics Model

**Solution
40 MHz**

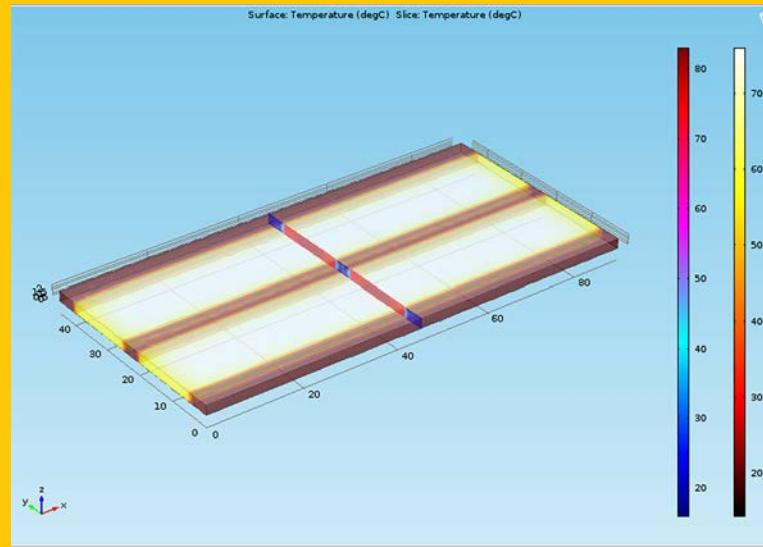


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Dielectric Heating: Solutions to the COMSOL Multiphysics Model

Solution
27 MHz

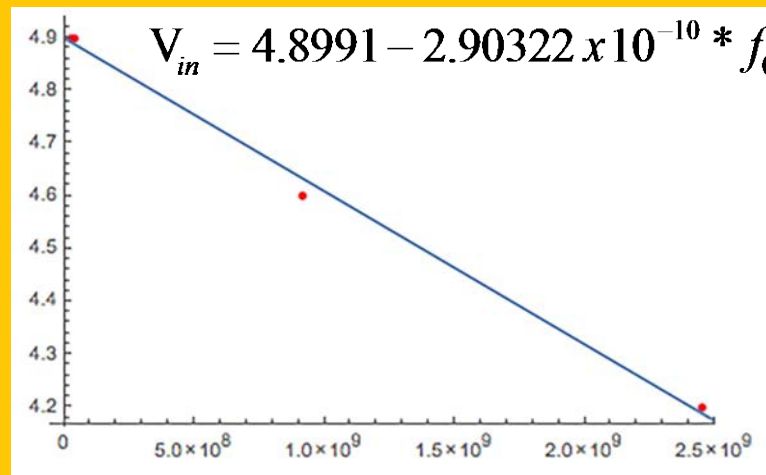


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Dielectric Heating: Solutions to the COMSOL Multiphysics Model

**Input
Voltage**



Frequency

Dielectric Heating: Conclusions

**The Heating Efficiency Increases
As
The Frequency Increases**

Thank You!